



with 20% piperidine in dimethylformamide. Cleavage from the resin was carried out with trifluoroacetic acid/anisole in both cases. Purification was done using chromatography on Sephadex G10 and preparative HPLC. The purity and the identity of the synthesized and purchased products were checked by analytical HPLC (better than 97% purity), NMR, FAB-MS, and amino acid analysis.

**Circular Dichroism Spectroscopy.** CD measurements were carried out on a Jasco J720 spectropolarimeter. All measurements were performed at room temperature. The path length of the quartz cell was either 1 or 0.1 mm. All spectra were corrected by subtracting the buffer baseline. Results are expressed in terms of mean residue ellipticity  $[\theta]$  in units of  $\text{deg cm}^2 \text{dmol}^{-1}$ . The percentages of the secondary structures were calculated with the protein secondary structure estimation program SSE-338 purchased from Jasco. This program is based on reference spectra of Yang et al. (1986).

**Sample Preparation.** CD measurements were performed at pH 4.0 and 5.5 using 5 mM sodium acetate buffer and at pH 7.4 using 5 mM MOPS. Appropriate amounts of buffer were added to the dry peptides and were equilibrated with stirring for 1–2 h. The concentration of peptide in the stock solution was 0.5 or 1 mM. For  $\beta\text{AP}(25\text{--}35)\text{OH}$  which was the least soluble of the three peptides, a 0.5 mM stock solution represents an equilibrium between colloidal fibrils and truly dissolved monomers. Stock solutions were freshly prepared for each set of dilution experiments. By monitoring the diluted sample with CD, a new equilibrium state was usually observed within 30 min. After equilibration the CD spectrum remained unchanged for 24 h ( $C_{\text{pep}} \leq 100 \mu\text{M}$ ) or at least a few hours (for higher peptide concentrations).

The peptide content of the stock solutions was determined by amino acid analysis.

**Molecular Modeling.** Modeling studies were performed using the Sybyl program package (Vers. 5.5. TRIPOS, St. Louis, MO). The starting structure was built using the ideal backbone torsion angles of an antiparallel  $\beta$ -pleated sheet for the whole sequence. Two molecules were adjusted to ensure the correct hydrogen bonding pattern for seven residues. The dimer was energy minimized to remove unfavorable interactions while keeping the hydrogen bonds and the salt bridge between the Lys-28 and Met-35 of opposite strands as constraints. The resulting structure must thus be considered as a local energy minimum.

**Ultracentrifugation Experiments.** Sedimentation equilibrium experiments were performed in a Beckman model XLA analytical ultracentrifuge. The 12-mm double-sector cell was filled with 0.12 mL solution in one sector and the same amount of solvent in the other (filling height  $\sim 3$  mm). Molecular weights were calculated using linear regression analysis to obtain the best linear fit of  $\ln(\text{absorbance})$  vs  $r^2$ . A partial specific volume of 0.73 mL/g was used.

## RESULTS

**Conformational Studies of  $\beta\text{AP}(25\text{--}35)\text{OH}$ .**  $\beta\text{AP}(25\text{--}35)\text{OH}$  is characterized by a free carboxy terminus at Met-35. The peptide conformation in solution was measured at pH 4.0, 5.5, and 7.4. Figure 1 displays CD spectra of  $\beta\text{AP}(25\text{--}35)\text{OH}$  at pH 4.0 measured as a function of peptide concentration. As the spectra were obtained by dilution of a concentrated stock solution we discuss them in the order of high concentration (curve 5)  $\rightarrow$  low concentration (curve 1). At high peptide concentrations, the CD spectra exhibit a distinct minimum at 217 nm which is the characteristic signature of the  $\beta$ -sheet structure. Upon dilution, the intensity

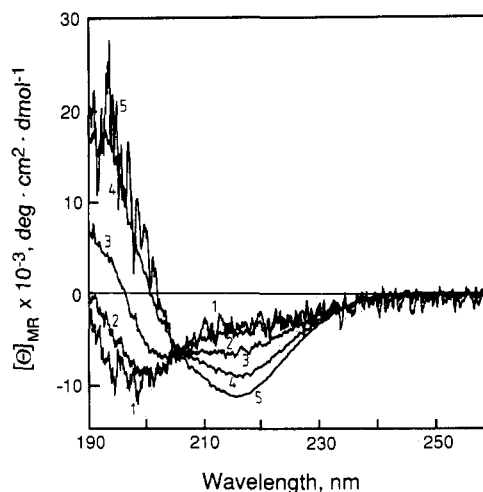


FIGURE 1: Circular dichroism spectra of  $\beta$ -amyloid peptide  $\beta\text{AP}(25\text{--}35)\text{OH}$  at pH 4.0 (5 mM sodium acetate buffer): concentration-dependent  $\beta$ -sheet  $\rightleftharpoons$  random coil transition. Random coil structure is favored at high dilution. Curve assignments are as follows: 1, 31; 2, 62.5; 3, 125; 4, 250; 5, 500  $\mu\text{M}$ .

of the 217-nm band is reduced and a new minimum develops at about 200 nm, indicating a predominantly random coil structure with some admixture of  $\beta$ -sheet structure. The  $\beta$ -sheet  $\rightarrow$  random coil transition exhibits an *isodichroic point* at 206 nm which provides evidence for a simple two-state  $\beta$ -sheet  $\rightleftharpoons$  random coil equilibrium. This is consistent with the absence of any notable  $\alpha$ -helix absorption at 208 and 222 nm.

Closer inspection of Figure 1 reveals a considerable increase in noise for the most concentrated solution (at short wavelength) indicative of enhanced light scattering. In order to detect the formation of peptide aggregates, the higher concentrated peptide solution (0.5 mM) was filtered through a Millipore filter of 0.22- $\mu\text{m}$  pore size. The filtrate was measured immediately after filtration and showed a predominantly random coil CD spectrum. Electron microscopy of the solid material retained by the millipore filter revealed small fibers.

The percentage of secondary structure was estimated with the computer program of the JASCO 720 spectropolarimeter, and the results are summarized in Table 1. Since the program is based on reference spectra obtained for proteins which differ in structure from amyloid proteins, the quantitative analysis must be considered as a first approximation only. Nevertheless, two features of this automatic fitting routine are obvious, namely (i) the percentage of random coil increases and (ii) the percentage of  $\beta$ -sheet decreases by a factor of 5–6 as the peptide concentration decreases from 500 to 10  $\mu\text{M}$ . The data thus demonstrate a *concentration-induced*  $\beta$ -sheet  $\rightleftharpoons$  random coil transition of  $\beta\text{AP}(25\text{--}35)\text{OH}$ .

Additional CD measurements on  $\beta\text{AP}(25\text{--}35)\text{OH}$  were performed at pH 5.5 and pH 7.4. At pH 5.5 the results were very similar to those obtained at pH 4.0: A concentration-dependent  $\beta$ -sheet  $\rightleftharpoons$  random coil transition was observed upon dilution of the stock solution (data not shown). In contrast, only a  $\beta$ -sheet structure was observed at pH 7.4, even at the lowest concentration measured (cf. Figure 2A). In addition, the most concentrated solution had a turbid appearance, indicating the formation of large peptide aggregates.

In the above dilution experiments CD spectra were measured typically 30 min after dilution of a freshly prepared stock solution. The CD spectra were unchanged when the same solutions were remeasured several hours later. However, a

Table 1. Conformational Analysis of the CD Spectra of  $\beta$ AP(25–35)OH as a Function of Peptide Concentration at pH = 4.0 in 5 mM AcONa Buffer

$C_{\text{pep}}$ ( $\mu\text{M}$ )	% random coil	% $\beta$ -sheet	% $\beta$ -turn	% $\alpha$ -helix
10	63.5	11.2	25.3	0
15.6	57.1	16.6	23.6	2.8
20	52.1	30.4	17.5	0
30	41.0	33.7	25.3	0
30	54.1	30.3	15.5	0
31	47.5	26.6	21.8	4.1
40	48.7	33.5	14.5	3.3
50	48.1	37.7	11.9	2.3
60	45.3	33.2	11.2	9.6
62.5	43.7	39	15.2	2.1
70	39.7	42.1	6.1	12.1
100	36.6	50.7	9.5	3.2
100	33	48.9	12.2	5.9
100	33.8	52.7	8.3	5.2
125	35.8	43.8	13.6	6.7
125	19.1	60.0	3.5	17.5
175	17.6	63.6	0.7	18.1
200	11.7	62.3	3.2	22.8
250	20.7	60.5	2.6	16.2
250	25.7	56.8	6.9	10.6
250	23.3	55	7.1	14.6
500	13.8	67.4	0	18.8
500	23	58.5	6.3	12.2
500	6.1	62.7	0	31.2
500	22.4	61	4.5	12.1

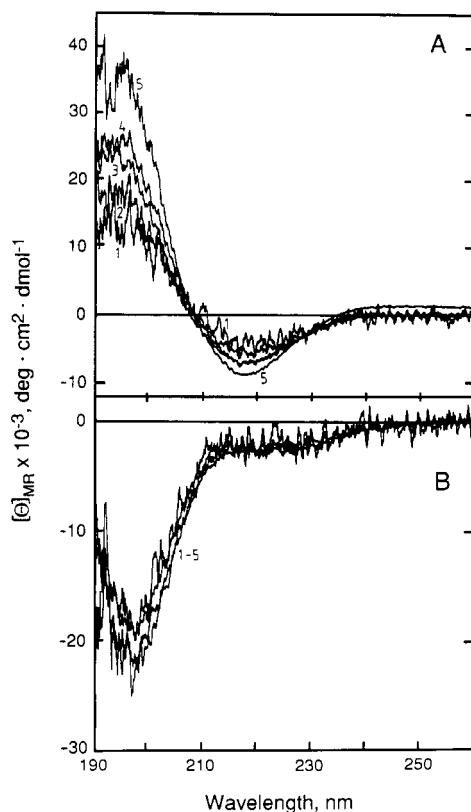


FIGURE 2: Comparison between fully charged  $\beta$ AP(25–35)OH and amidated  $\beta$ AP(25–35)NH<sub>2</sub>. Part A shows CD spectra of  $\beta$ AP(25–35)OH at pH 7.4 (5 mM MOPS). Independent of peptide concentration the peptide adopts predominantly a  $\beta$ -sheet structure. Curve assignments are as follows: 1, 31; 2, 63; 3, 100; 4, 250; 5, 375  $\mu\text{M}$ . Part B shows CD spectra of  $\beta$ AP(25–35)NH<sub>2</sub> at pH 4.0 (5 mM sodium acetate). The CD spectra are characteristic of a predominantly random coil structure at all concentrations. Curve assignments are as follows: 1, 35  $\mu\text{M}$ ; 2, 140  $\mu\text{M}$ ; 3, 333  $\mu\text{M}$ ; 4, 667  $\mu\text{M}$ ; 5, 1 mM.

different situation was encountered when the starting solution contained monomers only. This was achieved experimentally by separating fibrils from monomers using Millipore filters

(0.22- $\mu\text{m}$  diameter). For a solution with an initial peptide concentration of  $C_{\text{pep}} = 500 \mu\text{M}$  (at pH = 4), the corresponding filtrate revealed a random coil CD spectrum with a peptide concentration of  $C_{\text{pep}} \approx 129 \mu\text{M}$ , indicating a loss of 74% aggregated ( $\beta$ -sheet) peptide via filtration. For a 100  $\mu\text{M}$  solution, the filtrate had a concentration of  $\sim 45 \mu\text{M}$  and the loss was only 55%. Interestingly, the CD spectra of the filtrates revealed a random coil structure even 48 h after filtration. Apparently, the formation of new aggregates was kinetically hindered. Kinetic hindrance could however be relieved by seeding the solution with a few fibrils in agreement with recent experiments on related  $\beta$ APs and  $\beta$ AP fragments (Jarrett et al., 1993). Equilibration between random coil monomers and  $\beta$ -sheet fibrils was then established fairly rapidly as monitored with CD spectroscopy.

The monomeric state of  $\beta$ AP(25–35)OH in the filtered solutions was demonstrated with ultracentrifugation experiments. A 0.5 mM solution of  $\beta$ AP(25–35)OH in buffer (5 mM NaOAc, pH 4.0) was filtered through a 0.22- $\mu\text{m}$  Millipore filter and the filtrate centrifuged for 6 h at 56 000 rpm on a Beckman XLA analytical ultracentrifuge (An-60 Ti analytical rotor, 20 °C). The analysis of the sedimentation equilibrium (at 222 nm) yielded a molecular weight of  $1150 \pm 35$  Da (2 runs), in good agreement with the theoretical molecular weight of 1060.4 Da for  $\beta$ AP(25–35)OH monomers. No aggregates were detected. In contrast, when the measurement was repeated with an unfiltered sample, sedimentation of larger aggregates was observed starting at 10 000 rpm. Centrifugation was continued at 24 000 rpm until the aggregates reached the bottom. Increasing the rotational speed to 56 000 rpm led to the formation of a sedimentation equilibrium pattern characteristic of monomers only.

**Conformational Studies of  $\beta$ AP(25–35)NH<sub>2</sub> and  $\beta$ AP(25–35)NH<sub>2</sub>OH.** Peptide  $\beta$ AP(25–35)OH was modified by (i) substituting the carboxylate function by an amide function or (ii) by replacing methionine at position 35 by norleucine. For both peptides rather similar CD spectra were recorded at all concentrations and pH values. Figure 2B summarizes representative results for  $\beta$ AP(25–35)NH<sub>2</sub> at pH 4.0. The spectra are predominantly random coil (>60%) with some admixture of  $\beta$ -turn and  $\beta$ -sheet. For the norleucine derivative (data not shown), the percentage of random coil was reduced to 50% with a concomitant increase in  $\beta$ -turn and  $\beta$ -sheet. A reversible  $\beta$ -sheet  $\rightleftharpoons$  random coil transition was not observed. Changing the pH to 5.5 or 7.4 had no effect on the CD spectra of both peptides.

## DISCUSSION

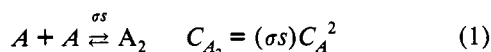
Human  $\beta$ -amyloid protein is a 39–43 amino acid polypeptide which is rather insoluble under physiological conditions. It has been proposed that the first 28 residues constitute an extracellular domain whereas residues 29–40 (43) are anchoring the peptide in the lipid membrane (cf. Fraser et al., 1991). The synthetic peptide  $\beta$ AP(25–35) overlaps partly with both domains. Lipid-binding studies revealed however no strong binding of  $\beta$ AP(25–35) to electrically neutral lipid bilayers (Terzi, E., Hölzemann, G., & Seelig, J., manuscript in preparation).

The most interesting property of  $\beta$ AP(25–35)OH is its reversible  $\beta$ -sheet  $\rightleftharpoons$  random coil transition at pH 4.0 and 5.5. Concentrated solutions of  $\beta$ AP(25–35)OH contain microfibrils with  $\beta$ -structure which upon dilution disintegrate into random coil monomers. Fibril formation from a purely monomeric solution appears to be kinetically hindered but can be catalyzed by seeding with  $\beta$ AP(25–35)OH fibrils.

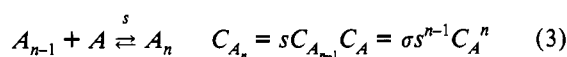
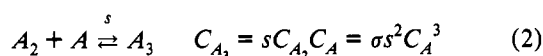
Human  $\beta$ AP(1–42) when dissolved in buffer containing 32.5% trifluoroethanol also exhibits a CD spectrum characteristic of  $\beta$ -sheet structure. However, dilution studies had practically no effect on the appearance of the CD spectrum (Otvos et al., 1993). In contrast, a  $\beta$ -sheet  $\rightleftharpoons$  random coil transition was observed for the related *rodent*  $\beta$ AP(1–42) which differs from *human*  $\beta$ AP(1–42) in three amino acid substitutions (Otvos et al., 1993). The dilution studies were again performed with a solvent containing 32.5% trifluoroethanol. TFE increases the solubility of the peptide and, at the same time, is thought to mimic the hydrophobic membrane environment.

Different fragments of  $\beta$ AP(1–42) have also been synthesized in order to elucidate the structural determinants involved in plaque formation. Both the proposed membrane-anchoring domain  $\beta$ AP(29–42) and the extracellular domain  $\beta$ AP(1–28) exhibit  $\beta$ -sheet and fibril formation in a narrow range of conditions (Halverson et al., 1990, 1991; Fraser et al., 1991; Barrow et al., 1992). None of these earlier studies on  $\beta$ AP fragments revealed a simple two-state equilibrium. We are also not aware of any quantitative treatment of the assembly process. Due to the reversibility of random coil  $\rightleftharpoons$   $\beta$ -sheet transition of  $\beta$ AP(25–35)OH, a simple thermodynamic analysis can now be given.

**Thermodynamic Analysis.** The CD spectra of  $\beta$ AP(25–35)OH measured as a function of concentration at pH 4.0 (Figure 1) and pH 5.5 showed an isodichroic point, which is consistent with the presence of just two conformations or two molecular species, i.e. monomers in the random coil conformation and oligomers with predominantly  $\beta$ -sheet structure. The concentration-dependent self-association can be described by a simple model which has been employed successfully in other aggregation processes of biomolecules (cf. Robinson et al., 1975; Heyn & Bretz, 1975; Cantor & Schimmel, 1980, p 145). The formation of a dimer is considered as the *nucleation step*



which is characterized by the equilibrium constant  $\sigma s$ .  $\sigma$  is denoted *nucleation parameter*. Fibers are formed by continuous addition of monomers according to the sequence



For the growth of the oligomers via the addition of monomers a constant *growth parameter*  $s$  was used. The total concentration of peptide,  $C_o$ , expressed in monomer units is given by

$$C_o = C_A + 2C_{A_2} + 3C_{A_3} + \dots + nC_{A_n}$$

$$C_o = C_A + 2\sigma s C_A^2 + \dots + n\sigma s^{n-1} C_A^n$$

$$C_o = C_A \{1 + \sigma(2sC_A + 3s^2 C_A^2 + \dots + ns^{n-1} C_A^{n-1})\} \quad (4)$$

The evaluation of the latter expression is straightforward with the assumptions of  $n \rightarrow \infty$  and  $sC_A < 1$  leading to

$$C_o = C_A \left\{ 1 - \sigma + \frac{\sigma}{(1 - sC_A)^2} \right\} \quad (5)$$

For a non-cooperative model, i.e.  $\sigma = 1$ , eq 5 is further

simplified to

$$C_o = C_A / (1 - sC_A)^2 \quad (6)$$

For a given set of parameters  $\sigma$ ,  $s$ , and  $C_o$  eqs 5 and 6 allow the evaluation of the monomer concentration  $C_A$ . The concentration of monomers forming aggregates is then given by  $C_\beta = C_o - C_A$  and the *mole fraction* of monomers involved in  $\beta$ -structured fibers is found as

$$X_\beta = (C_o - C_A) / C_o \quad (7)$$

Table 1 summarizes the quantitative evaluation of the CD spectra of  $\beta$ AP(25–35)OH in terms of percentages of random coil,  $\beta$ -turn, and  $\beta$ -sheet structures. Molecules with  $\beta$ -sheet structure are involved in fibril formation, whereas those adopting a random coil or  $\beta$ -turn structure are assumed to occur as monomers only. The mole fraction of aggregated monomers involved in fibril formation, denoted  $X_\beta$  in the following, can be calculated as

$$X_\beta = \frac{\% \beta\text{-structure observed}}{\text{maximum } \% \beta\text{-structure}}$$

For the maximum amount of  $\beta$ -structure we assumed 70%  $\beta$ -structure (which is close to the experimental result observed at the highest peptide concentration). Figure 3 displays the calculated  $X_\beta$  values and also the best theoretical fits using the cooperative and the non-cooperative models. The theory suggests a cooperativity parameter of  $\sigma = 0.2$  and a binding constant of  $s = (2.9 \pm 1.2) \times 10^4 \text{ M}^{-1}$  for the cooperative model and  $s = (1.8 \pm 1) \times 10^4 \text{ M}^{-1}$  for the non-cooperative model ( $\sigma = 1$ ). The free energy of association is  $\Delta G \approx -6.0$  kcal/mol for both models. In addition, within the accuracy of the measurement, the data obtained at pH 5.5 lead to the same thermodynamic parameters as those measured at pH 4.0.

In order to estimate the enthalpy of association,  $\Delta H_a$ , dilution experiments were performed using titration calorimetry (E. Terzi, G. Hölzemann, and J. Seelig, manuscript in preparation). A concentrated solution of  $\beta$ AP(25–35)OH was injected into a pure buffer solution. An *endothermic* dissociation reaction of about +3 kcal/mol was observed, i.e. the association reaction is *exothermic* by the same amount. The enthalpy of association,  $\Delta H_a \approx -3$  kcal/mol and thus accounts for at least 50% of the free energy of aggregation.

The carboxy terminus of  $\beta$ AP(25–35)OH appears to play an important role in  $\beta$ -sheet formation. If the carboxylic group is *fully* charged (pH 7.4) the CD spectra reflect maximum  $\beta$ -structure at all concentrations (Figure 2A). For a *partially* charged carboxy terminus (pH 4) a concentration-dependent  $\beta$ -sheet  $\rightleftharpoons$  random coil transition is observed. *Elimination* of charge by amidation of the carboxy terminus leads to a predominantly random coil structure (Figure 2B). A possible interpretation of these results is suggested by Figure 4 which depicts two molecules of  $\beta$ AP(25–35) aligned in an antiparallel  $\beta$ -pleated sheet. The dimer was energy minimized to remove all unfavorable interactions thus showing a local energy minimum structure (cf. Materials and Methods).

The characteristic features of this model are two intermolecular ion pairs between Lys-28 and Met-35, with 3 amino acids of each peptide dangling at the end of the  $\beta$ -sheet. This model suggests a maximum of 70%  $\beta$ -structure, in agreement with the experimental observations. Ionic interactions then appear to play a central role in the stability of this  $\beta$ -sheet. Protonation of the carboxyl group or removal of its electric charge by amidation shifts the equilibrium toward the random coil monomeric state. However, if Met-35 is replaced by Nle,

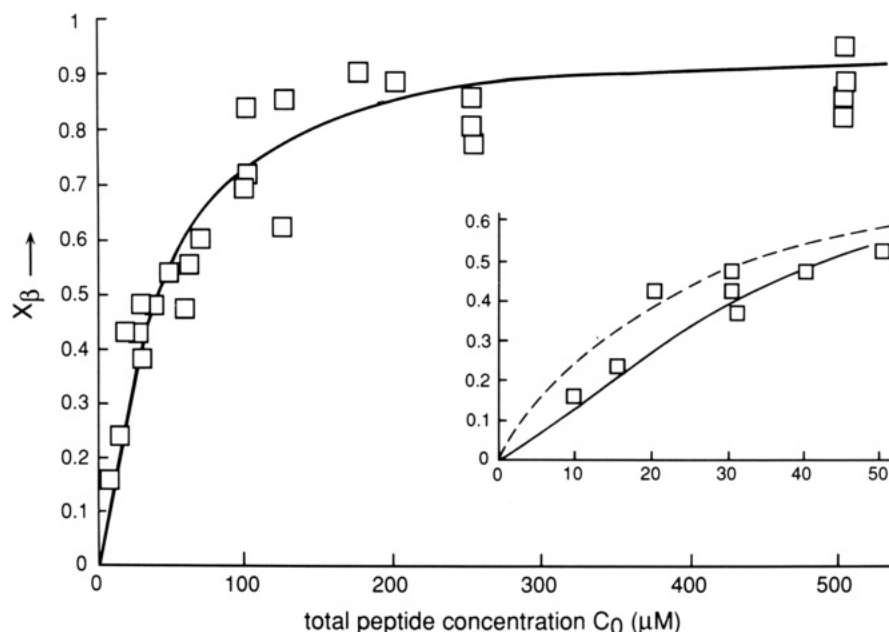


FIGURE 3: Concentration-dependent random coil  $\rightleftharpoons$   $\beta$ -sheet transition of  $\beta$ AP(25–35)OH at pH 4.0. The mole fraction of  $\beta$ -sheet,  $X_\beta$  (i.e. the mole fraction of monomers involved in  $\beta$ -sheet formation), is plotted versus the total monomer concentration. The experimental results are compared with a cooperative model (—,  $\sigma = 0.2$ ;  $s = 2.9 \times 10^4 \text{ M}^{-1}$ ) and a non-cooperative model (---,  $\sigma = 1$ ;  $s = 1.8 \times 10^4 \text{ M}^{-1}$ ). The difference between the two models is most pronounced in the low concentration range.

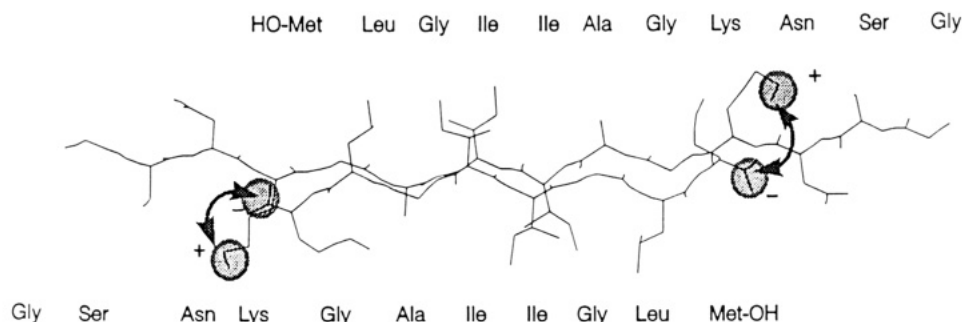


FIGURE 4: Molecular model of  $\beta$ -sheet formation by  $\beta$ AP(25–35)OH. Two molecules of  $\beta$ AP(25–35)OH are aligned antiparallel to each other. Ion pair interactions stabilize the two ends of the  $\beta$ -pleated sheet structure.

the amidated derivative ( $\beta$ AP(25–35Nle)NH<sub>2</sub>) undergoes a concentration-dependent random coil  $\rightleftharpoons$   $\beta$ -structured aggregate transition (E. Terzi, G. Hölzemann, and J. Seelig, manuscript in preparation). Nle is a well-known  $\beta$ -structure promoter and this result suggests that the ionic interaction Lys-28...Met-35 is not an absolute requirement for  $\beta$ -sheet formation but can be replaced by a strong van der Waals interaction.

$\beta$ AP(1–43) and some of its fragments can adopt quite different conformations depending on the solution conditions (cf. Barrow et al., 1992). Using solvent mixtures with fluorinated alcohols such as trifluoroethanol it has been possible, for example, to induce an  $\alpha$ -helix  $\rightleftharpoons$   $\beta$ -sheet transition in the  $\beta$ AP(1–42) and  $\beta$ AP(1–39) peptides (Barrow et al., 1992). It has further been demonstrated that fibril assembly and disassembly are pH-dependent processes, with  $\beta$ -structure formation being favored at pH 4 to 7 (Fraser et al., 1991; Burdick et al., 1992; Barrow & Zagorski, 1991; Barrow et al., 1992). The present study then appears to be the first quantitative description of a reversible, concentration-induced random coil  $\rightleftharpoons$   $\beta$ -sheet equilibrium. Moreover, fragment  $\beta$ AP-(25–35)OH is one of the shortest peptide molecules observed so far to exhibit the formation of large  $\beta$ -structured fibrils. Segment  $\beta$ AP(25–35)OH could thus be involved in the initial interactions leading to the aggregation of the full-length peptide  $\beta$ AP(1–43) in amyloid-like deposits.

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## REFERENCES

- Barrow, C. J., & Zagorski, M. G. (1991) *Science* 253, 179–182.
- Barrow, C. J., Yasuda, A., Kenny, P. T. M., & Zagorski, M. G. (1992) *J. Mol. Biol.* 225, 1075–1093.
- Burdick, D., Soreghan, B., Kwon, M., Kosmoski, J., Knauer, M., Henschen, A., Yates, J., Cotman, C., & Glabe, C. (1992) *J. Biol. Chem.* 267, 546–554.
- Burgevin, M. C., Daniel, N., Doble, A., & Blanchard, J. C. (1992) *Neuroreport* 3, 1131–1134.
- Cantor, C. R., & Schimmel, P. R. (1980) *Biophysical Chemistry*, Vol. 1, p 145, Freeman, San Francisco.
- Cheung, N. S., Small, D. H., & Livett, B. G. (1993) *J. Neurochem.* 60, 1163–1166.
- Distefano, M., Aleppo, G., Casabona, G., Genazzani, A. A., Scapagnini, U., & Nicoletti, F. (1993) *Brain Res.* 600, 166–168.
- Fraser, P. E., Nguyen, J. T., Surewicz, W. K., & Kirschner, D. A. (1991) *Biophys. J.* 60, 1190–1201.
- Glenner, G. G., & Wong, C. W. (1984) *Biochem. Biophys. Res. Commun.* 120, 885–890.

- Halverson, K., Fraser, P. E., Kirschner, D. A., & Lansbury, P. T. (1990) *Biochemistry* 29, 2639–2644.
- Halverson, K., Sucholeiki, I., Ashburn, T. T., & Lansbury, P. T. (1991) *J. Am. Chem. Soc.* 113, 6701–6703.
- Heyn, M. P., & Bretz, R. (1975) *Biophys. Chem.* 3, 35–45.
- Jarrett, J. T., Berger, E. P., & Landsbury, P. T. (1993) *Biochemistry* 32, 4693–4697.
- Kang, J., Lemaire, H.-G., Unterbeck, A., Salbaum, J. M., Masters, C. L., Grzeschik, K.-H., Multhaup, G., Beyreuther, K., & Müller-Hill, B. (1987) *Nature* 325, 733–736.
- Lansbury, P. T. (1992) *Biochemistry* 31, 6865–6870.
- Lee, J. M., Weinstein, D. A., Kowall, N. W., & Beal, M. F. (1992) *Drug Dev. Res.* 27, 441–444.
- Masters, C. L., Multhaup, G., Simms, G., Pottgiesser, J., Martins, R. N., & Beyreuther, K. (1985a) *EMBO J.* 4, 2757–2763.
- Masters, C. L., Simms, G., Weinman, N. A., Multhaup, G., Simms, G., McDonald, B. L., & Beyreuther, K. (1985b) *Proc. Natl. Acad. Sci. U.S.A.* 82, 4245–4249.
- Otvos, L., Szendrei, G. I., Lee, V. M. Y., & Mantsch, H. H. (1993) *Eur. J. Biochem.* 211, 249–257.
- Pike, C. J., Burdick, D., Walencewicz, A. J., Glabe, C. G., & Cotman, C. W. (1993) *J. Neurosci.* 13, 1676–1687.
- Rink, H. (1987) *Tetrahedron Lett.* 28, 3787–3790.
- Robinson, B. H., Seelig, A., & Schwarz, G. (1975) in *Chemical and Biological Applications of Relaxation Spectrometry* (Wyn-Jones, H., Ed.) pp 481–485, Reidel, Dordrecht, Holland.
- Yang, J. T., Wu, C.-S. C., & Martinez, H. M. (1986) *Methods Enzymol.* 130, 208–269.
- Yankner, B. A., Duffy, L. K., & Kirschner, D. A. (1990) *Science* 250, 279–282.